I Number	Hits	Search Text	DB	Time stamp
L Number	260706	polysaccharide cellulose starch pecti\$2	USPAT;	2003/04/11 11:55
	200700	chitin chitosan polyglucosamine hyaluron\$4	US-PGPUB	2003,01,11 11.33
2	326364		USPAT; US-PGPUB	2003/04/11 11:05
3	9855	(polysaccharide cellulose starch pecti\$2 chitin chitosan polyglucosamine hyaluron\$4 chondroitin) same (oxidati\$4 oxidiz\$4)	USPAT; US-PGPUB	2003/04/11 09:52
4	23046	hypochlorous hypochlorite tempo nitroxyl tetramethylpiperidine	USPAT; US-PGPUB	2003/04/11 11:06
5	6089	(oxidati\$4 oxidiz\$4) same (hypochlorous hypochlorite tempo nitroxyl tetramethylpiperidine)	USPAT; US-PGPUB	2003/04/11 09:55
6	810	((polysacharide cellulose starch pecti\$2 chitin chitosan polyglucosamine hyaluron\$4 chondroitin) same (oxidati\$4 oxidiz\$4)) and ((oxidati\$4 oxidiz\$4) same (hypochlorous hypochlorite tempo nitroxyl tetramethylpiperidine))	USPAT; US-PGPUB	2003/04/11 09:55
7	494834	alkali merceriz\$6 gelatiniz\$6 wettable wettabil\$6 solubil\$6 soluble	USPAT; US-PGPUB	2003/04/11 09:56
8	309	(hypochlorous hypochlorite) and (tempo nitroxyl tetramethylpiperidine)	USPAT; US-PGPUB	2003/04/11 09:57
9	32	((polysacharide cellulose starch pecti\$2 chitin chitosan polyglucosamine hyaluron\$4 chondroitin) same (oxidati\$4 oxidiz\$4)) and ((hypochlorous hypochlorite) and (tempo nitroxyl tetramethylpiperidine))	USPAT; US-PGPUB	2003/04/11 09:57
10	1245028	ph mw (mol adj (wt weight)) (molecular weight)	USPAT; US-PGPUB	2003/04/11 12:00
11	29	(oxidati\$4 oxidiz\$4) and (alkali merceriz\$6 gelatiniz\$6 wettable wettabil\$6 solubil\$6 soluble) and (((polysaccharide cellulose starch pecti\$2 chitin chitosan polyglucosamine hyaluron\$4 chondroitin) same (oxidati\$4 oxidiz\$4)) and ((hypochlorous hypochlorite) and (tempo nitroxyl tetramethylpiperidine))) and (ph mw (mol adj (wt weight)) (molecular weight))	USPAT; US-PGPUB	2003/04/11 11:04
12	179294	polysaccharide cellulose starch pecti\$2 chitin chitosan polyglucosamine hyaluron\$4 chondroitin	EPO; JPO; DERWENT	2003/04/11 11:05
13	243685	oxidati\$4 oxidiz\$4	EPO; JPO; DERWENT	2003/04/11 11:05
14	3792	<pre>(polysaccharide cellulose starch pecti\$2 chitin chitosan polyglucosamine hyaluron\$4 chondroitin) and (oxidati\$4 oxidiz\$4)</pre>	ÉPO; JPO; DERWENT	2003/04/11 11:05
15	11840	hypochlorous hypochlorite tempo nitroxyl tetramethylpiperidine	EPO; JPO; DERWENT	2003/04/11 11:06
16	159	((polysaccharide cellulose starch pecti\$2 chitin chitosan polyglucosamine hyaluron\$4 chondroitin) and (oxidati\$4 oxidiz\$4)) and (hypochlorous hypochlorite tempo nitroxyl tetramethylpiperidine)	EPO; JPO; DERWENT	2003/04/11 11:07
17	15402	chitin chitosan polyglucosamine hyaluron\$4 chondroitin	USPAT; US-PGPUB	2003/04/11 11:55
18	590	(oxidati\$4 oxidiz\$4) same (chitin chitosan polyglucosamine hyaluron\$4 chondroitin)	USPAT; US-PGPUB	2003/04/11 11:55
19	495494	absorb\$6 superabsorb\$6	USPAT; US-PGPUB	2003/04/11 11:57
20	341	((oxidati\$4 oxidiz\$4) same (chitin chitosan polyglucosamine hyaluron\$4 chondroitin)) and (absorb\$6 superabsorb\$6)	USPAT; US-PGPUB	2003/04/11 11:57
21	41	<pre>((oxidati\$4 oxidiz\$4) same (chitin chitosan polyglucosamine hyaluron\$4 chondroitin)) same (absorb\$6 superabsorb\$6)</pre>	USPAT; US-PGPUB	2003/04/11 11:59
22	246487	carboxyl\$6	USPAT; US-PGPUB	2003/04/11 12:00

23	366	((oxidati\$4 oxidiz\$4) same (chitin	USPAT:	2003/04/11 12:00
		chitosan polyglucosamine hyaluron\$4	US-PGPUB	
	• %	chondroitin)) and carboxy1\$6		
24	300918	mw (mol adj (wt weight)) (molecular adj	USPAT;	2003/04/11 12:01
	+	weight)	US-PGPUB	
25	281	(((oxidati\$4 oxidiz\$4) same (chitin	USPAT;	2003/04/11 12:01
		chitosan polyglucosamine hyaluron\$4	US-PGPUB	
,		chondroitin)) and carboxyl\$6) and (mw		
		(mol adj (wt weight)) (molecular adj		
		weight))		
26	281	1 (USPAT;	2003/04/11 12:02
		polyglucosamine hyaluron\$4 chondroitin)	US-PGPUB	
		and ((((oxidati\$4 oxidiz\$4) same (chitin		
		chitosan polyglucosamine hyaluron\$4	ł	-
		chondroitin)) and carboxyl\$6) and (mw		1
		(mol adj (wt weight)) (molecular adj		
		weight)))		

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ANSWER 1 OF 33 CAPLUS COPYRIGHT 2003 ACS
                          2003:58309
ACCESSION NUMBER:
DOCUMENT NUMBER:
                          138:108525
                          Cationic cellulosic fibers for paper making
TITLE:
                          Van Brussel-Verraest, Dorine Lisa; Besemer, Arie
INVENTOR(S):
                          Cornelis; Thiewes, Harm Jan; Verwilligen,
                          Anne-Mieke Yvonne Wilhelmina
PATENT ASSIGNEE(S):
                          SCA Hygiene Products Zeist B.V., Neth.; SCA Hygiene
                          Products AB
SOURCE:
                          PCT Int. Appl., 16 pp.
                          CODEN: PIXXD2
DOCUMENT TYPE:
                          Patent
LANGUAGE:
                          English
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                                             APPLICATION NO. DATE
     PATENT NO.
                      KIND DATE
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                             -----
                                             _____
                                             WO 2002-NL458
     WO 2003006739
                             20030123
                                                               20020711
                       A1
         W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN,
             CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR,
             LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH,
             PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU,
             TJ, TM
         RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AT, BE, BG,
             CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL,
             PT, SE, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG
                      A1
     US 2003024662
                             20030206
                                             US 2002-192691
                                                               20020711
     US 2003026828
                       A1
                             20030206
                                             US 2002-192715
                                                               20020711
                                          EP 2001-202667 A 20010711
PRIORITY APPLN. INFO.:
                                          EP 2001-205070 A 20011221
US 2001-304111P P 20010711
                                          US 2001-341856P P 20011221
     A cationic cellulosic fiber contg. between 1 and 30 cationic groups and
     between 0.1 and 20 aldehyde groups per 100 anhydroglucose units is a
     suitable basis for producing paper and tissue products without the
     necessity of using non-biodegradable cationic polymers as wet strength
     additives. The cationic cellulosic fiber can be obtained by oxidn
     . of the fiber to introduce aldehyde groups, followed by reaction of part
     of the aldehyde groups with a nitrogen-contg. reagent such as betaine
     hydrazide hydrochloride. The fiber is advantageously combined with an
     anionic polymer such as monoaldehyde carboxyl-starch or with anionic
     cvclodextrin.
REFERENCE COUNT:
                                THERE ARE 2 CITED REFERENCES AVAILABLE FOR THIS
                          2
                              RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT
    ANSWER 2 OF 33 CAPLUS COPYRIGHT 2003 ACS
ACCESSION NUMBER:
                          2003:40129 CAPLUS
DOCUMENT NUMBER:
                          138:95664
                          Odor regulation in hygiene products
TITLE:
INVENTOR(S):
                          Besemer, Arie Cornelis; Thiewes, Harm Jan
PATENT ASSIGNEE(S):
                          SCA Hygiene Products A.B., Swed.
SOURCE:
                          Eur. Pat. Appl., 10 pp.
                          CODEN: EPXXDW
DOCUMENT TYPE:
                          Patent
LANGUAGE:
                          English
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
     PATENT NO. KIND DATE
                                            APPLICATION NO. DATE
     EP 1275404 A1 20030115
                                             _____
                                            EP 2001-202668 20010711
         R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
             IE, SI, LT, LV, FI, RO, MK, CY, AL, TR
     WO 2003006078
                       A2
                            20030123
                                             WO 2002-NL456
                                                               20020711
         W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN,
             CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH,
             GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR,
             LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH,
             PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TN, TR, TT, TZ,
             UA, UG, US, UZ, VN, YU, ZA, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU,
             TJ, TM
         RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL,
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PT, SE, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR,

NE, SN, TD, TG

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US 2003086974 A1 20030508
                                              US 2002-192690
                                                                 20020711
PRIORITY APPLN. INFO.:
                                          EP 2001-202668 A 20010711
US 2001-304408P P 20010712
     Oxidized cyclodextrins contg. an av. of at least 0.5 carboxyl
     group and/or at least 0.5 functional group selected from aldehyde groups
      and aldehyde-derived groups per mol. can be used as complexing agents, in
      particular in odor control for hygiene products. A fragrance component
      can be incorporated in the oxidized cyclodextrin and released
      when in use, and, in addn. or alternatively, the oxidized
      cyclodextrin can absorb undesired odor components during use.
                                 THERE ARE 8 CITED REFERENCES AVAILABLE FOR THIS
REFERENCE COUNT:
                           8
                                 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT
     ANSWER 3 OF 33 CAPLUS COPYRIGHT 2003 ACS
                          2002:811998 CAPLUS
ACCESSION NUMBER:
DOCUMENT NUMBER:
                           137:312634
                           Process for oxidizing primary hydroxyls in
TITLE:
                           carbohydrates
                           Besemer, Arie; Van Brussel-Verraest, Dorine
INVENTOR(S):
                           Lisa; Thiewes, Harm Jan
PATENT ASSIGNEE(S):
                           SCA Hygiene Products Zeist B.V., Neth.
                           Eur. Pat. Appl., 6 pp.
SOURCE:
                           CODEN: EPXXDW
DOCUMENT TYPE:
                           Patent
LANGUAGE:
                           English
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:
     PATENT NO.
                       KIND DATE
                                             APPLICATION NO. DATE
     EP 1251140 A1 20021023 EP 2001-201454 20010420
         R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR
PRIORITY APPLN. INFO.:
                                          EP 2001-201454
     Carbohydrates having primary hydroxyl groups, such as starch and
     cellulose, can be selectively oxidized to carboxylic
     carbohydrates by oxidn. with mol. oxygen, using a nitrogen oxide
     such as a nitrite salt, as a catalyst. The reaction is advantageously
     carried out in a dehydrating solvent such as concd. phosphoric acid.
     Thus, adding 500 .mu.L 65% nitric acid and 50 mg Na nitrite to a dissoln.
     of 2.5 g potato starch (15.4 mmol anhydroglucose units) in 50 mL of 85%
     H3PO4, exposing the resulting mixt. to O and oxidizing gave
     oxidized starch with degree of oxidn. (Blumenkrantz
method) 40 and 50% after 24 and 96 h, resp.
REFERENCE COUNT:
                                 THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS
                           3
                                 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT
    ANSWER 4 OF 33 CAPLUS COPYRIGHT 2003 ACS
ACCESSION NUMBER:
                           2002:575029 CAPLUS
DOCUMENT NUMBER:
                           137:124781
TITLE:
                           Recovery of nitroxyl radicals from oxidation
                           reactions
INVENTOR(S):
                           Thornton, Jeff; Besemer, Arie; Schraven, Bas
PATENT ASSIGNEE(S):
                           SCA Hygiene Products AB, Swed.
SOURCE:
                           PCT Int. Appl., 22 pp.
                           CODEN: PIXXD2
DOCUMENT TYPE:
                           Patent
LANGUAGE:
                           English
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:
     PATENT NO.
                      KIND DATE
                                             APPLICATION NO. DATE
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     WO 2002059064
                      A1 20020801
                                              WO 2001-SE2632
                                                                20011129
         W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN,
              CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR,
              LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH,
              PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM
         RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AT, BE, CH,
              CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG
     SE 2001000210
                     A 20020727
A1 20021017
                              20020727
                                              SE 2001-210
                                                                20010126
     US 2002151431
                                              US 2002-53646
                                                                20020124
                                           SE 2001-210 A 20010126
US 2001-264018P P 20010126
PRIORITY APPLN. INFO.:
                                           WO 2001-SE2632 W 20011129
                          CASREACT 137:124781
OTHER SOURCE(S):
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AB Stable nitroxyl radicals, such as TEMPO and its derivs., used as catalysts

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in oxidn. reactions are recovered from oxidn.
     reactions by hydrophobic interactions with polymers, such as XAD resins,
     .beta.-cyclodextrin or silica gel. Thus, potato starch in water was
     treated with 4-acetamido-TEMPO and NaOCl at pH 8.5-9.5. The reaction
     mixt. was run through a column of silica gel, eluted with water. The
     6-carboxy starch was eluted first, followed by the 4-acetamido-TEMPO which
     could be recycled without loss of activity.
                                   THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS
REFERENCE COUNT:
                                   RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT
    ANSWER 5 OF 33 CAPLUS COPYRIGHT 2003 ACS
                            2002:505677 CAPLUS
ACCESSION NUMBER:
DOCUMENT NUMBER:
                            137:247354
TITLE:
                            Selective oxidation of carbohydrates by
                            4-AcNH-TEMPO/peracid systems
                            Bragd, Petter L.; Besemer, Arie C.; Van
AUTHOR (S):
                            Bekkum, Herman
CORPORATE SOURCE:
                            SCA Hygiene Products, Zeist, 3700 AJ, Neth.
                            Carbohydrate Polymers (2002), 49(4), 397-406
SOURCE:
                            CODEN: CAPOD8; ISSN: 0144-8617
PUBLISHER:
                            Elsevier Science Ltd.
DOCUMENT TYPE:
                            Journal
LANGUAGE:
                            English
    Starch, amylopectin, inulin, pullulan and Me .alpha.-D-glucopyranoside (Me
      .alpha.-Glcp) were oxidized by 4-acetamido-2,2,6,6-
     tetramethylpiperidine-1-oxyl (4-AcNH-TEMPO) as the mediator and peracetic
     acid or monoperoxysulfate (Oxone) as the regenerating oxidant. The
     conversion of primary alc. groups to the corresponding carboxyl groups proceeded with high yield and selectivity, provided that NaBr was added as
     co-catalyst. The mass mol. distributions of the oxidized
     polysaccharides indicated that no major depolymn. occurred during
     oxidn. Oxone appeared to be the most efficient oxidant as the
     reaction rate was 25 times higher than that of peracetic acid in the
     oxidn. of Me .alpha.-Glcp. However, oxone produces a larger amt.
     of waste as byproduct than peracetic acid.
REFERENCE COUNT:
                            50
                                   THERE ARE 50 CITED REFERENCES AVAILABLE FOR THIS
                                   RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT
L5 ANSWER 6 OF 33 CAPLUS COPYRIGHT 2003 ACS
                            2002:462498 CAPLUS
ACCESSION NUMBER:
DOCUMENT NUMBER:
                            137:21652
                            Production of high molecular weight oxidized
TITLE:
                            cellulose
INVENTOR (S):
                            Besemer, Arie Cornelis; Van
Brussel-Verraest, Dorine Lisa
PATENT ASSIGNEE(S):
                            SCA Hygiene Products Zeist B.V., Neth.
SOURCE:
                            Eur. Pat. Appl., 6 pp.
                            CODEN: EPXXDW
DOCUMENT TYPE:
                            Patent
LANGUAGE .
                            English
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                    KIND DATE
     PATENT NO.
                                                APPLICATION NO. DATE
                               -----
                        ----
     EP 1215217 A1 20020619
                                               EP 2000-204465 20001212
         R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR
                                                WO 2001-NL902
     WO 2002048196
                         A1 20020620
                                                                    20011212
          W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN,
              CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH,
              GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH,
              \mathtt{PL},\ \mathtt{PT},\ \mathtt{RO},\ \mathtt{RU},\ \mathtt{SD},\ \mathtt{SE},\ \mathtt{SG},\ \mathtt{SI},\ \mathtt{SK},\ \mathtt{SL},\ \mathtt{TJ},\ \mathtt{TM},\ \mathtt{TN},\ \mathtt{TR},\ \mathtt{TT},\ \mathtt{TZ},
          UA, UG, US, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AT, BE, CH,
              CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG
     AU 2002019717
                        A5 20020624
                                               AU 2002-19717
                                                                    20011212
PRIORITY APPLN. INFO.:
                                             EP 2000-204465 A
                                                                    20001212
                                                                W 20011212
                                             WO 2001-NL902
     The invention pertains to a process for producing oxidized
     cellulose having a high mol. wt., by pre-treating cellulose in a non-alk.
     solvent for cellulose so as to lower its crystallinity, and then
     oxidizing it using an oxidizing system predominantly
     oxidizing the 6-hydroxymethyl groups or the 2,3-dihydroxyethylene
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groups to carbaldehyde and/or carboxyl groups. The oxidizing agent is preferably a nitroxyl compd. (TEMPO) or periodate, and the

acid-regenerated cellulose (2 g) was suspended in water, TEMPO (40 mg) and

product is useful as a water-absorbent. Thus, a phosphoric

NaBr (1 g) were added. Sodium hypochlorite (2 M) was added in 2 mL portions to a total of 12 mL. The pH was kept at 10.5 by addn. of 0.5 M NaOH (total addn. 20.3 mL). The temp. was kept at 4.degree. during the reaction. The total reaction time was 5.5 h. The oxidized product was completely water-sol. Na borohydride (100 mg) was added to reduce aldehyde groups. The product was then isolated by pptn., washed and dried under vacuum at room temp. (yield 2.2 g). The oxidn. degree of the product (based on sodium hydroxide consumption) was 85%. The product consisted of one monodisperse fraction with an av. mol. wt. of 410,000. The oxidized cellulose was dissolved in water (10% soln.), the pH was adjusted to 4.5 and butanediol diglycidyl ether was added (10 mol%). The crosslinking was carried out at 50 degree. for 20 h. The obtained gel was dried at 100.degree. in a fluidized bed dryer, reswollen in excess water and dried again. The particles were ground to 100-800 .mu.m particles. The absorption under load in synthetic urine was 11 g/g. Crosslinking with divinyl sulfone resulted in even better performance than with butanediol diglycidyl ether.

THERE ARE 6 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: 6 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L5 ANSWER 7 OF 33 CAPLUS COPYRIGHT 2003 ACS 2002:450359 CAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 137:21693

Process for oxidizing primary alcohols such TITLE:

as that of polysaccharides

INVENTOR (S) · Bragd, Petter; Besemer, Arie Cornelis

PATENT ASSIGNEE(S): Swed.

U.S. Pat. Appl. Publ., 3 pp. SOURCE:

CODEN: USXXCO

DOCUMENT TYPE: Patent

LANGUAGE: English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

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PATENT NO.
                              KIND DATE
                                                            APPLICATION NO. DATE
                              ----
                               A1 20020011
A1 20020620
       US 2002072600
                                       20020613
                                                            US 2001-13654
                                                                                     20011213
      WO 2002048197
                                                            WO 2001-NL903
                                                                                     20011212
            W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN,
                  CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR,
                  LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH,
                  PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM
            RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG
      AU 2002019718
                              A5 20020624
                                                            AU 2002-19718
                                                                                     20011212
                                                        EP 2000-204483 A 20001213
US 2000-255899P P 20001218
PRIORITY APPLN. INFO.:
                                                        WO 2001-NL903
                                                                             W 20011212
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Primary hydroxyl groups in a substrate having both primary and secondary AB hydroxyl groups can be selectively oxidized to carbaldehyde and/or carboxyl groups by contacting the substrate with a cyclic nitroxyl compd. in the presence of a peroxosulfate as a co-oxidant and by carrying out the reaction at a temp. below 30.degree. and at a pH below 9. The process is halogen-free and metal-free and is esp. suitable for oxidizing polysaccharides. Thus, 3.0 g potato starch was gelatinized in 200 mL deionized water at 95.degree. with effective mech. stirring and to which 61 mg 4-acetamido-TEMPO was added and dissolved. The soln. was cooled on ice and the temp. was maintained at .ltoreq.10.degree. throughout the reaction. The reaction was initiated by the addn. of 11.38 g Oxone (i.e., 2 mol HSO5-/mol primary alc.), which was added under mech. stirring in small portions throughout the oxidn . to minimize unwanted side reactions. After each addn. of the acidic oxidant, pH was raised to 8.2 and then kept const. by the addn. of 0.5 M NaOH using a pH stat app. After completion of the reaction (8-10 h), the remaining aldehyde intermediates were reduced to the starting alc. using 150 mg of NaBH4. After one hour, pH was adjusted to .apprx.6.0 with 0.5 M HCl and the reacted polysaccharides were pptd. in 2 vol. of ethanol, filtered off, and re-dissolved in 50 mL of water. Finally the materials were freeze-dried. The conversion of the primary hydroxys to carboxys was estd. by 13CNMR to be 60 mol%.

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ANSWER 8 OF 33 CAPLUS COPYRIGHT 2003 ACS
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ACCESSION NUMBER: 2002:450358 CAPLUS

DOCUMENT NUMBER: 137:34718

Recovery of spent periodate in oxidation of TITLE:

starch

INVENTOR(S): Besemer, Arie

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PATENT ASSIGNEE(S):
                     Neth.
                     U.S. Pat. Appl. Publ., 4 pp.
SOURCE:
                     CODEN: USXXCO
DOCUMENT TYPE:
                     Patent
                     English
LANGUAGE:
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:
                  KIND DATE
                                   APPLICATION NO. DATE
    PATENT NO.
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US 2002072599 A1 20020613 US 2001-13449 20011213 WO 2002048029 A1 20020620 WO 2001-NL904 20011212 W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG AU 2002-19719 20011212 EP 2000-204482 A 20001213 US 2000-255898P P 20001218 WO 2001-NL904 W 20011212 AU 2002019719 A5 20020624 PRIORITY APPLN. INFO.:

Spent periodate, used in the oxidn. of starch, was regenerated and recovered by reacting the periodate with at least an equimolar amt. of peroxosulfate. Thus, a periodate soln. was used to oxidize potato starch to dialdehyde starch and treated with Oxone to give periodate suitable for further oxidn. of starch.

ANSWER 9 OF 33 CAPLUS COPYRIGHT 2003 ACS ACCESSION NUMBER: 2002:450357 CAPLUS

DOCUMENT NUMBER: 137:21651

Production of high molecular weight oxidized TITLE:

cellulose

INVENTOR (S): Besemer, Arie Cornellis; Van Brussel-Verraest, Dorine Lisa PATENT ASSIGNEE(S):

Neth.

SOURCE: U.S. Pat. Appl. Publ., 4 pp.

CODEN: USXXCO

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE ----US 2002072598 A1 20020613 US 2001-11976 20011211 US 2000-254587P P 20001212 PRIORITY APPLN. INFO.: AB The invention pertains to a process for producing oxidized cellulose having a high mol. wt., by pre-treating cellulose in a non-alk. solvent for cellulose so as to lower its crystallinity, and then oxidizing it using an oxidizing system predominantly oxidizing the 6-hydroxymethyl groups or the 2,3-dihydroxyethylene groups to carbaldehyde and/or carboxyl groups. The oxidizing agent is preferably a nitroxyl compd. (TEMPO) or periodate, and the product is useful as a water absorbent. Thus, a phosphoric acid-regenerated cellulose (2 g) was suspended in water to which TEMPO (40 mg) and NaBr (1 g) were added. Sodium hypochlorite (2 M) was added in 2 mL portions to a total of 12 mL. The pH was kept at 10.5 by addn. of 0.5 M NaOH (total addn. 20.3 mL). The temp. was kept at 4.degree. during the reaction. The total reaction time was 5.5 h. The oxidized product was completely water-sol. Na borohydride (100 mg) was added to reduce aldehyde groups. The product was then isolated by pptn. and dried. The oxidn. degree of the product (based on NaOH consumption) was 85%. The product consisted of one monodisperse fraction with an av. mol. wt. of 410,000. The oxidized cellulose was dissolved in water (10% soln.), the pH was adjusted to 4.5 and butanediol diglycidyl ether was added (10 mol%). The crosslinking was carried out at 50.degree. for 20 h. The obtained gel was dried at 100.degree. in a fluidized bed dryer, reswollen in excess water and dried again. The particles were ground to 100-800 .mu.m particles. The absorption under load in synthetic urine was 11 g/g. Crosslinking with divinyl sulfone resulted in even better performance than with butanediol diglycidyl ether.

ANSWER 10 OF 33 CAPLUS COPYRIGHT 2003 ACS ACCESSION NUMBER: 2001:817022 CAPLUS

DOCUMENT NUMBER: 135:359319

TITLE: Aldehyde-containing polymers as wet strength additives in paper making

INVENTOR(S):

Thornton, Jeffrey Wilson; Van Brussel-Verraest, Dorine

Lisa; Besemer, Arie; Sandberg, Sussan Sca Hygiene Products Zeist B.V., Neth.

PATENT ASSIGNEE(S):

PCT Int. Appl., 24 pp.

SOURCE:

CODEN: PIXXD2

DOCUMENT TYPE:

Patent English

LANGUAGE:

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

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PATENT NO.
                       KIND DATE
                                                APPLICATION NO. DATE
                                                WO 2001-NL343
                                                                   20010504
     WO 2001083887
                              20011108
                         A1
          W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN,
              CO, CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM,
              HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS,
              LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO,
              RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM
          RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY,
              DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
                        A1 20011114
                                               EP 2000-201693 20000511
     EP 1154074
          R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
              IE, SI, LT, LV, FI, RO
                        A1 20020307
A1 20030129
     US 2002026993
                                                US 2001-848212
                                                                   20010504
                                               EP 2001-928253 20010504
     EP 1278913
          R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
              IE, SI, LT, LV, FI, RO, MK, CY, AL, TR
PRIORITY APPLN. INFO.:
                                            EP 2000-201630 A 20000504
                                            EP 2000-201693 A 20000511
US 2000-203913P P 20000512
                                            WO 2001-NL343
                                                               W 20010504
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Water-sol. or water-dispersible polymers (and cationic derivs.) having a mol. wt. .gtoreq.800, contg. .gtoreq.5 aldehyde groups/mol. and .gtoreq.1 AB carboxyl group/mol., the ratio of aldehyde groups to carboxyl groups being >0.75:1, are used as a wet strength additive in paper making. dialdehyde starch (100% oxidized) was prepd. and slurried in 100 mL water and tissue hand sheets using 10 mg/g additive and 10 mg/g polyamide polyimide epichlorohydrin adduct, showing wet strength 3.2 Nm/g; vs. 2.5 Nm/g using polyamide polyimide epichlorohydrin adduct only. THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT:

ANSWER 11 OF 33 CAPLUS COPYRIGHT 2003 ACS

ACCESSION NUMBER:

2001:781472 CAPLUS

DOCUMENT NUMBER:

135:332682

TITLE:

Oxidation of polysaccharides with nitroxyls

RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

INVENTOR(S): Bragd, Petter; Besemer, Arie Cornelis;

Thornton, Jeffrey Wilson Neth.

PATENT ASSIGNEE(S):

SOURCE:

U.S. Pat. Appl. Publ., 3 pp.

CODEN: USXXCO

DOCUMENT TYPE:

Patent English

LANGUAGE:

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND DATE	APPLICATION NO.	DATE		
US 2001034442	A1 20011025	US 2001-841083	20010425		
EP 1149846	A1 20011031	EP 2000-201461	20000425		
R: AT, BE,	CH, DE, DK, ES,	FR, GB, GR, IT, LI, LU,	NL, SE, MC, PT,		
IE, SI,	LT, LV, FI, RO				
JP 2002003503	A2 20020109	JP 2001-125591	20010424		
PRIORITY APPLN. INFO.	:	EP 2000-201461 A	20000425		
AB A simplified pro	cess for oxidizi	ng starch and other			
polysaccharides :	in an aq. soln.	or suspension using hyp	ochlorite in the		
presence of a catalytic amt. of a nitroxyl compd. (e.q.,					
2,2,6,6-tetramethylpiperidine-1-oxyl) is described. The oxidn.					
process is bromide-free and is carried out at a pH between 7 and 9.3 and					
at a temp. betwee	en 15 and 25.deg	ree			

ANSWER 12 OF 33 CAPLUS COPYRIGHT 2003 ACS ACCESSION NUMBER: 2001:360047 CAPLUS

DOCUMENT NUMBER:

134:354734

TITLE:

Oxidized polysaccharides and products made

thereof

INVENTOR(S):

Jaschinski, Thomas; Gunnars, Susanna; Besemer,

Arie Cornelis; Bragd, Petter

PATENT ASSIGNEE(S):

SCA Hygiene Products G.m.b.H., Germany

PCT Int. Appl., 51 pp.

DOCUMENT TYPE:

CODEN: PIXXD2

Patent

LANGUAGE:

English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

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APPLICATION NO. DATE
PATENT NO.
                    KIND DATE
WO 2001034656
                     A1
                            20010517
                                              WO 2000-EP11048 20001108
     W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN,
         CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT,
          LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU,
         SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM
     RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF,
          BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
                     A1
                            20010523
                                               DE 1999-19953589 19991108
DE 19953589
                                              BR 2000-15245
                            20020723
                                                                    20001108
BR 2000015245
                     Α
                                              EP 2000-972899 20001108
EP 1228099
                     A1
                           20020807
    R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, MC, IE, SI,
         LT, LV, FI, RO, MK, CY, AL
                                          JP 2001-537367 20001108
DE 1999-19953589 A 19991108
JP 2003514077
                     T2 20030415
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PRIORITY APPLN. INFO.: WO 2000-EP11048 W 20001108

The present invention relates to a polysaccharide having functional groups, wherein said groups are aldehyde groups formed at positions C2 and/or C3 as well as at position C6 of the anhydroglucose units of the polysaccharide chain. Preferably, the polysaccharide is a cellulosic fibrous material, the primary and secondary hydroxyl groups of which are at least partially oxidized to aldehyde groups by means of TEMPO oxidn. and periodate oxidn. The invention also relates

to a paper or nonwoven comprising the above polysaccharide. According to the invention a relative wet strength of greater than 10% can be achieved. THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

ANSWER 13 OF 33 CAPLUS COPYRIGHT 2003 ACS ACCESSION NUMBER: 2001:330920 CAPLUS 135:122663

DOCUMENT NUMBER:

TITLE: TEMPO-derivatives as catalysts in the oxidation of primary alcohol groups in

carbohydrates

AUTHOR (S) . Bragd, Petter L.; Besemer, Arie C.; van

Bekkum, Herman

CORPORATE SOURCE: SCA Hygiene Products, Zeist, 3704 AJ, Neth.

Journal of Molecular Catalysis A: Chemical (2001), SOURCE:

170(1-2), 35-42

CODEN: JMCCF2; ISSN: 1381-1169

PUBLISHER: Elsevier Science B.V.

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 135:122663

Primary hydroxyl groups in aq. starch, pullulan and Me .alpha.-D-glucopyranoside were oxidized to the corresponding carboxylic acid functionalities by TEMPO-(4-X)-derivs. using sodium hypochlorite as the primary oxidant. All the combinations of substrates and nitroxyl radicals resulted in stoichiometric conversions, and the selectivity for oxidn. of primary hydroxyls was high. Some depolymn. occurred throughout the oxidn., esp. when 4-acetoxy and 4-mesyl-TEMPO were used. The pH window of the activity of the inexpensive 4-acetamido-TEMPO was found to be substantially lower from that of the other tested TEMPO-derivs.; thus allowing milder reaction conditions. At pH 8, the rate of oxidn. was ca. two times higher when 4-acetamido-TEMPO was used compared to the other catalysts.

THERE ARE 24 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: 24 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

ANSWER 14 OF 33 CAPLUS COPYRIGHT 2003 ACS

ACCESSION NUMBER:

2001:29875 CAPLUS

DOCUMENT NUMBER:

134:71837

TITLE: Process for regioselective oxidation of

primary alcohols of carbohydrates in preparation of

uronic acids

INVENTOR(S): Besemer, Arie Cornelis; Jaschinski, Thomas PATENT ASSIGNEE(S): SCA Hygiene Products Zeist B.V., Neth.

PCT Int. Appl., 11 pp. SOURCE: CODEN: PIXXD2 DOCUMENT TYPE: Patent English LANGUAGE: FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION: PATENT NO. KIND DATE APPLICATION NO. DATE WO 2001000681 A1 20010104 Al 20010104 WO 2000-NL453 20000628 W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM

RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG A1 20010103 EP 1999-202126 19990630 EP 1065218 R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO
DE 10084757 T 20020829 DE 2000-10084757 20000628 EP 1999-202126 A 19990630 WO 2000-NL453 W 20000628 PRIORITY APPLN. INFO.: WO 2000-NL453 OTHER SOURCE(S): CASREACT 134:71837 The invention concerns a process for oxidizing a primary alc. such as a carbohydrate using an oxidizing agent in the presence of a catalytic amt. of a di-tertiary-alkyl nitroxyl, wherein the alc. is oxidized using an oxidic compd. of a period 4 or 5 metal having an oxidn. state of a least +3 as an oxidizing agent, for example manganese dioxide. REFERENCE COUNT: 5、 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT ANSWER 15 OF 33 CAPLUS COPYRIGHT 2003 ACS ACCESSION NUMBER: 2000:881088 CAPLUS DOCUMENT NUMBER: 134:44078 Process for regenerating periodic acid TITLE: INVENTOR(S): Besemer, Arie Cornelis; Jetten, Jan Mattijs PATENT ASSIGNEE(S): Sca Hygiene Products Zeist B.V., Neth. PCT Int. Appl., 10 pp. CODEN: PIXXD2 DOCUMENT TYPE: Patent English LANGUAGE: FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION: APPLICATION NO. DATE 2000075070 Al 20001214 WO 2000-NL386 20000607 W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CR, WO 2000075070 CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG BR 2000011371 A 20020226 BR 2000-11371 20000607 EP 1189834 A1 20020327 EP 2000-939190 20000607 R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO
JP 2003501334 T2 20030114
US 6538132 B1 20030325 JP 2001-501556 20000607

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US 6538132

B1 20030325

US 2001-980789

20011206

PRIORITY APPLN. INFO.:

EP 1999-201808

A 19990607

WO 2000-NL386

W 20000607

AB Periodic acid is regenerated and recovered from a spent iodate soln. by reaction with at least an equimolar amt. of a hypohalite in the presence of a water-miscible org. solvent, K+ or divalent cations (esp. Ca2+, Mg2+). The periodic acid is suitable for oxidn. of carbohydrates to dialdehyde carbohydrates, e.g., starch to dialdehyde starch, a wet strength additive for paper. Dialdehyde starch can be further oxidized to dicarboxy starch, a sequestering agent.

REFERENCE COUNT:

5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT
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Process for selective oxidation of cellulose
TITLE:
INVENTOR(S):
                            Jetten, Jan Matthijs; Van Den Dool, Ronald; Van
                            Hartingsveldt, Wim; Besemer, Arie Cornelis
                            Sca Hygiene Products Zeist B.V., Neth.
PATENT ASSIGNEE(S):
SOURCE:
                            PCT Int. Appl., 14 pp.
                            CODEN: PIXXD2
DOCUMENT TYPE:
                            Patent
LANGUAGE:
                            English
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
     PATENT NO.
                                               APPLICATION NO. DATE
                        KIND DATE
     WO 2000050463
                         A1
                               20000831
                                                WO 2000-NL119
                                                                   20000224
         W: AE, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CR, CU,
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              MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI,
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         155039 A1 20011121 EP 2000-906770 20000224 R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
     EP 1155039
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                               20020205
                                                BR 2000-8475
                                                                   20000224
                                               JP 2000-601041 20000224
     JP 2002537448
                         T2
                               20021105
                                             EP 1999-200537 A 19990224
WO 2000-NL119 W 20000224
PRIORITY APPLN. INFO.:
   A process for oxidizing cellulose, in which a nitroxyl compd.
     such as 2,2,6,6-tetramethylpiperidin-1-oxy (TEMPO) is oxidized
     using an oxidizing agent in the presence of a complex of a
     transition metal such as Mn, Fe, Cu, and a complexing agent such as a
     polyamine, or an oxidative enzyme, and the resulting nitrosonium
     ion is used to selectively oxidize the cellulose
     6-hydroxy-methylene groups to carbaldehyde groups and carboxylic acid
     groups.
REFERENCE COUNT:
                            10
                                  THERE ARE 10 CITED REFERENCES AVAILABLE FOR THIS
                                  RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT
    ANSWER 17 OF 33 CAPLUS COPYRIGHT 2003 ACS
ACCESSION NUMBER:
                            2000:608782 CAPLUS
DOCUMENT NUMBER:
                            133:209532
TITLE:
                            Oxidized cellulose-containing fibrous
                            materials, preparation thereof and products therefrom
INVENTOR(S):
                            Jaschinski, Thomas; Gunnars, Susanna; Besemer,
                            Arie Cornelis; Bragd, Petter; Jetten, Jan
                            Matthijs; Van den Dool, Ronald; Van Hartingsveldt,
                            Willem
PATENT ASSIGNEE(S):
                            Sca Hygiene Products G.m.b.H., Germany; Sca Hygiene
                            Products Zeist B.V.
                            PCT Int. Appl., 75 pp.
SOURCE:
                            CODEN: PIXXD2
DOCUMENT TYPE:
                            Patent
LANGUAGE:
                            English
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
     PATENT NO.
                        KIND DATE
                                                APPLICATION NO. DATE
     WO 2000050462
                               20000831
                                                WO 2000-EP1538 20000224
         W: AE, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CR, CU, CZ, DE, DK, DM, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL,
              IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW, AM,
              AZ, BY, KG, KZ, MD, RU, TJ, TM
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              DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF,
              CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
                       A1 20010517
A1 20011121
                                               DE 1999-19953590 19991108
     DE 19953590
     EP 1155040
                                                EP 2000-907622 20000224
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                                                BR 2000-8378
                                                                   20000224
     JP 2002537503
                         T2
                               20021105
                                                JP 2000-601040
                                                                   20000224
     US 2002098317
                         A1 20020725
                                               US 2001-931621
                                                                   20010816
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EP 1999-200537 A 19990224

DOCUMENT NUMBER:

PRIORITY APPLN. INFO.:

133:209514

DE 1999-19953590 A 19991108 WO 2000-EP1538 W 20000224

A cellulose-contg. fibrous material is prepd. by oxidizing hydroxy groups at the C(6) of glucose units of cellulose into aldehyde and/or carboxy groups, and used to prep. paper or nonwoven products, esp. tissue products. The paper or nonwoven products display excellent strength properties. Thus, bleached hardwood sulfite pulp was treated for 60 min under acidic conditions with 4-hydroxy-TEMPO (1.00 g/50 g dry fibrous material) using 5% of 13% NaOCl as a primary oxidizing agent, and used to prep. test sheets (basis wt. 80 g/m2) having wt. 2.56 g, breaking strength 23.94 (dry) and 4.687 N/15 mm (wet), tear length 1980.1 (dry) and 387.7 m (wet), and rel. wet strength 19.6%, compared with 3.04, 18.48, 0.151, 1285.7, 10.5, and 0.8, resp., for a nonoxidized pulp. THERE ARE 8 CITED REFERENCES AVAILABLE FOR THIS REFERENCE COUNT: 8 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT ANSWER 18 OF 33 CAPLUS COPYRIGHT 2003 ACS 2000:314401 CAPLUS ACCESSION NUMBER: DOCUMENT NUMBER: 132:323226 TITLE: Carbohydrate oxidation products INVENTOR (S): Besemer, Arie Cornelis; Jetten, Jan Matthijs; Van Doren, Hendrik Arend; Van Der Lugt, Jan Pieter Nederlandse Organisatie voor Toegepast-PATENT ASSIGNEE(S): Natuurwetenschappelijk Onderzoek TNO, Neth. Eur. Pat. Appl., 7 pp. SOURCE: CODEN: EPXXDW DOCUMENT TYPE: Patent LANGUAGE: English FAMILY ACC. NUM. COUNT: PATENT INFORMATION: PATENT NO. KIND DATE APPLICATION NO. DATE EP 999222 A1 20000510 EP 1998-203706 19981102 R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO 2000026257 Al 20000511 WO 1999-NL673 19991102 W: AE, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CR, CU, WO 2000026257 CZ, DE, DK, DM, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM RW: GH, GM, KE, LS, MW, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG EP 1999-971429 19991102 EP 1137672 A1 20011004 R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO JP 2002528605 T2 20020903 JP 2000-5/5043 = 19981102 EP 1998-203706 A 19981102 WO 1999-NI673 W 19991102 JP 2000-579643 19991102 PRIORITY APPLN. INFO.: A novel oxidn. product derived from a carbohydrate contg. 1,2-dihydroxyethylene groups in its repeating units, can be obtained by at least partially oxidizing the carbohydrate 1,2-dihydroxyethylene groups to dialdehyde groups, and oxidizing a part of the aldehyde groups to CO2H groups. The oxidn. product has a regular structure with alternating CHO groups and CO2H groups in a ratio of about 1:1. It can be further transformed to an amino-substituted oxidn. product by reductive amination of at least a part of the remaining CHO groups. Thus, starch was oxidized with NaIO4 and the resulting starch dialdehyde further oxidized with Br (in situ generated from NaBr and AcOOH) to give starch comprising 0.7 CO2H and 1.2 CHO groups per monomeric unit. REFERENCE COUNT: R THERE ARE 8 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT ANSWER 19 OF 33 CAPLUS COPYRIGHT 2003 ACS SSION NUMBER: 1999:736773 CAPLUS ACCESSION NUMBER: DOCUMENT NUMBER: 131:352732 TITLE: Amino-carboxylic acid derivatives of carbohydrates as chelating agents INVENTOR(S): Van Brussel-Verraest, Dorine L.; Besemer, Arie C.; Thornton, Jeffrey W. PATENT ASSIGNEE(S): SCA Hygiene Products Nederland B.V., Neth. PCT Int. Appl., 17 pp. CODEN: PIXXD2

DOCUMENT TYPE:

LANGUAGE:

Patent

English

FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION:

PATENT NO.

KIND DATE

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APPLICATION NO. DATE
                                -----
                        ____
                                                  WO 1999-NL300
      WO 9958574
                         A1
                               19991118
                                                                     19990517
          W: AE, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ,
               DE, DK, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS,
               JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ,
               TM, TR, TT, UA, UG, US, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ,
          MD, RU, TJ, TM
RW: GH, GM, KE, LS, MW, SD, SL, SZ, UG, ZW, AT, BE, CH, CY, DE, DK,
               ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
      AU 9941720
                         A1 19991129
                                                 AU 1999-41720
                                                                      19990517
PRIORITY APPLN. INFO.:
                                               EP 1998-201586
                                                                      19980514
                                              WO 1999-NL300
                                                                      19990517
      The title derivs. are prepd., wherein .gtoreq.1 -CHOH or -CH2OH group per
      10 monosaccharide units is converted to a group having the formula
      -CH2A[NH(CH2)nCHR1CO]mOH, wherein m=1-10, n=0-4, A=a direct bond or a (poly) aminoalkylene group, R1 = H, carboxyl, or C1-C4 alkyl optionally
      substituted by OH, MeO, SH, MeS, substituted mercapto or dithio, amino,
      guanidino, guanyl, ureido, carboxyl, carbamoyl, Ph, substituted Ph or a heterocyclic group, or, if n .noteq. 0, R1 may also be an amino. Other
      both carboxylated and aminated carbohydrates are equiv. to the derivs.
      defined above. These derivs. are useful for binding transition metals,
     e.g. before bleaching of pulp. Thus, a chelating agent was prepd. from oxidized starch, aspartic acid, and Na cyanoborohydride.
                                    THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS
REFERENCE COUNT:
                            4
                                    RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT
     ANSWER 20 OF 33 CAPLUS COPYRIGHT 2003 ACS
                             1999:723070 CAPLUS
ACCESSION NUMBER:
DOCUMENT NUMBER:
                             131:337312
TITLE:
                             Process for selective oxidation of primary
                             alcohols
INVENTOR(S):
                             Van Der Lugt, Jan Pieter; Jetten, Jan Matthijs;
                             Besemer, Arie Cornelis; Van Doren, Hendrik
                             Arend
PATENT ASSIGNEE(S):
                             Nederlandse Organisatie voor Toegepast
                             Natuurwetenschappelijk Onderzoek TNO, Neth.
SOURCE:
                             PCT Int. Appl., 11 pp.
                             CODEN: PIXXD2
DOCUMENT TYPE:
                             Patent
LANGUAGE:
                             English
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
      PATENT NO.
                         KIND DATE
                                                 APPLICATION NO. DATE
                         ----
                                                  -----
      WO 9957158
                                                 WO 1999-NL272
                          A1
                                19991111
                                                                     19990504
          W: AE, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ,
               DE, DK, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS,
               JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK,
               MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, UA, UG, US, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ,
               MD, RU, TJ, TM
          RW: GH, GM, KE, LS, MW, SD, SL, SZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG,
               CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
     CA 2331701
                          AA 19991111
                                                  CA 1999-2331701 19990504
     AU 9937369
                          A1
                                19991123
                                                 AU 1999-37369
                                                                     19990504
     AU 746462
                          B2
                                20020502
     BR 9910274
                          Α
                                20010102
                                                 BR 1999-10274
                                                                     19990504
     EP 1093467
                          A1
                                20010425
                                                 EP 1999-919713 19990504
                          B1
     EP 1093467
                                20020327
          R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, NL, SE, PT, IE, FI
     AT 215097
                          E
                                20020415
                                                 AT 1999-919713
                                                                     19990504
     JP 2002513809
                          T2
                                20020514
                                                  JP 2000-547126
                                                                     19990504
     NZ 507873
                          Α
                                20020531
                                                  NZ 1999-507873
                                                                     19990504
     ES 2174605
                          Т3
                                20021101
                                                 ES 1999-919713
                                                                     19990504
                          B1 20030211
     US 6518419
                                                 US 2000-706767
                                                                     20001107
PRIORITY APPLN. INFO.:
                                              EP 1998-201495 A 19980507
                                              WO 1999-NL272
                                                                  W 19990504
OTHER SOURCE(S):
                           CASREACT 131:337312
     Primary alcs., esp. in carbohydrates, can be selectively oxidized to aldehydes and carboxylic acids in a low-halogen process by using a
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peracid in the presence of a catalytic amt. of a di-tertiary-alkyl

nitroxyl (TEMPO) and a catalytic amt. of halide. The halide is preferably

bromide and the process can be carried out at nearly neutral to moderately alk. pH (5-11). The peracid can be produced or regenerated by means of hydrogen peroxide or oxygen. The process is advantageous for producing uronic acids and for introducing aldehyde groups which are suitable for crosslinking and derivatization. THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS 5

REFERENCE COUNT: RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

ANSWER 21 OF 33 CAPLUS COPYRIGHT 2003 ACS ACCESSION NUMBER: 1999:375255 CAPLUS

DOCUMENT NUMBER: 131:23575

TITLE: Superabsorbent material made from oxidized

polysaccharides

Besemer, Arie Cornelis; Thornton, Jeffrey INVENTOR (S):

Wilson

PATENT ASSIGNEE(S): SCA Molnlycke, Neth. Eur. Pat. Appl., 7 pp. SOURCE:

CODEN: EPXXDW

DOCUMENT TYPE:

Patent

LANGUAGE:

English FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE ----EP 1997-203823 19971205 EP 920874 A1 19990609 R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO WO 1998-NL693 WO 9929352 A1 19990617 19981207 W: AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ, DE, DK, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, UA, UG, US, UZ, VN, YU, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM RW: GH, GM, KE, LS, MW, SD, SZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG A1 19990628 AU 9915122 AU 1999-15122 19981207 EP 1997-203823 PRIORITY APPLN. INFO.: 19971205

WO 1998-NL693 19981207 The invention provides a process of producing a superabsorbent polysaccharide deriv. by oxidn. of a .alpha.-glucan, glucomannan or galactomannan to introduce aldehyde functions; the oxidized polysaccharide subsequently being reacted with sulfur dioxide or an equiv. thereof to produce a hydroxysulfonated polysaccharide. The polysaccharide may be crosslinked before or after the oxidn. The preferred polysaccharide is starch or guar. Thus, 10 g starch was suspended in 14% sodium sulfate soln. and was crosslinked with 0.1% epichlorohydrin. After 20 h the product was isolated and washed, then treated with 96% ethanol and acetone and dried in vacuum oven. The dried material was suspended in 250 mL water and 6.6 g sodium periodate was added and the mixt. was stirred for 20 h to obtain an oxidized product with a 50% degree of oxidn. The product was collected by filtration washed, dried, and freeze-dried. To a suspension of 1 g of the dry material in 20 mL of water, 1.6 mL of 39% sodium bisulfite soln. was added to obtain a gel which was pptd. in 96% ethanol and then collected and dried. The product had a free swelling capacity of 1.0 g/g.

REFERENCE COUNT: 4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

ANSWER 22 OF 33 CAPLUS COPYRIGHT 2003 ACS 1999:222866 CAPLUS

ACCESSION NUMBER:

DOCUMENT NUMBER: 130:224921

Cleaning of water filters with calcium-binding agents TITLE:

and catalytic oxidation in the presence of a

di-tert-nitroxyl compound

INVENTOR(S): Besemer, Arie Cornelis; Jetten, Jan

Matthijs; Broens, Lute

PATENT ASSIGNEE(S): Norit Membraan Technologie B.V., Neth.

SOURCE: PCT Int. Appl., 17 pp.

CODEN: PIXXD2

DOCUMENT TYPE:

Patent English

LANGUAGE: FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE WO 9915256 A1 19990401 WO 1998-NL544 19980921

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W: AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ, DE, DK, EE, ES, FI, GB, GE, GH, GM, HR, HU, ID, IL, IS, JP, KE, KG,
              KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX,
              NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT,
              UA, UG, US, UZ, VN, YU, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM
          RW: GH, GM, KE, LS, MW, SD, SZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI,
              CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
     NL 1007086
                       C2
                               19990322
                                               NL 1997-1007086 19970919
     AU 9891901
                                               AU 1998-91901
                         A1
                               19990412
                                                                   19980921
PRIORITY APPLN. INFO.:
                                            NL 1997-1007086
                                                                   19970919
                                            WO 1998-NL544
                                                                   19980921
     Filters for water purifn. can be cleaned by treatment with a
     calcium-binding agent, preferably followed by catalytic oxidn., for example with hypochlorite in the presence of 2,2,6,6-
     tetramethylpiperidine-N-oxyl or a similar nitroxyl. Another oxidn
     . system is with H2O2 in the presence of a transition metal complex with a
     cyclic polyamine. The calcium-binding agent can be a
     dicarboxypolysaccharide, a phosphate, or a polyacylamine (e.g., EDTA or
     NTA). In particular, dicarboxyinulin, optionally in combination with a
     zeolite, can be used as calcium-binding agent.
                                  THERE ARE 6 CITED REFERENCES AVAILABLE FOR THIS
REFERENCE COUNT:
                           6
                                  RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT
    ANSWER 23 OF 33 CAPLUS COPYRIGHT 2003 ACS
                           1998:26714 CAPLUS
ACCESSION NUMBER:
DOCUMENT NUMBER:
                            128:115171
TITLE:
                           Autocatalytic oxidation of primary hydroxyl
                            functions in glucans with nitrogen oxides
AUTHOR (S):
                           de Nooy, Arjan E. J.; Pagliaro, Mario; van Bekkum,
                           Herman; Besemer, Arie C.
                           Department of Biochemistry, TNO Nutrition and Food Research Institute, Zeist, 3700 AJ, Neth.
CORPORATE SOURCE:
SOURCE:
                           Carbohydrate Research (1997), 304(2), 117-123
                           CODEN: CRBRAT; ISSN: 0008-6215
                           Elsevier Science Ltd.
PUBLISHER:
DOCUMENT TYPE:
                           Journal
LANGUAGE:
                           English
     The selective oxidn. of the primary hydroxyl groups in the
     glucans cellulose, amylose and pullulan with nitrogen oxides has been
     studied. The polymers were dissolved in 85% phosphoric acid and sodium
     nitrate was used as the stoichiometric oxidant. A catalytic amt. of sodium nitrite was added to reduce the induction time. With this reaction
     system, where the oxidizing nitrogen oxides are formed in situ,
     the primary hydroxyl groups could be completely oxidized (> 95%)
     to carboxylic acids. Undesired ketones due to secondary hydroxyl group
     oxidn. were subsequently reduced with sodium borohydride. Esp.
     for the .alpha.-glucans, this oxidn.-redn. sequence of secondary
     hydroxyl functions apparently gave epimerization. Degrdn. of the polymers
     was slow provided the oxidn. was performed at 4 .degree.C.
     Thus, pullulan with <Mw> .apprxeq. 170 kg/mol yielded a polyuronate with
     <Mw> .apprxeq. 100 kg/mol. A study of this reaction system with
     .beta.-cyclodextrin as the substrate clearly showed that the reaction was
     autocatalytic.
     ANSWER 24 OF 33 CAPLUS COPYRIGHT 2003 ACS
ACCESSION NUMBER:
                           1996:646015 CAPLUS
DOCUMENT NUMBER:
                           125:300167
TITLE:
                           On the use of stable organic nitroxyl radicals for the
                           oxidation of primary and secondary alcohols
AUTHOR (S):
                           De Nooy, Arjan E. J.; Besemer, Arie C.; Van
                           Bekkum, Herman
CORPORATE SOURCE:
                           Department Biochemistry, TNO Nutrition Food Research
                           Institute, Zeist, NL 3700, Neth.
Synthesis (1996), (10), 1153-1174
SOURCE:
                           CODEN: SYNTBF; ISSN: 0039-7881
PUBLISHER:
                           Thieme
DOCUMENT TYPE:
                           Journal; General Review
LANGUAGE:
                           English
     A review with >111 refs. on the oxidn. of primary and secondary
     alcs. with nitroxyl radicals. A variety of stoichiometric and catalytic
     methods is discussed, with reactions in org. solvents, under biphasic
     conditions, and in water. Mechanistic studies and general exptl. procedures are included. The high regionelectivity for the oxidn
      of primary alcs. that can be achieved in the presence of secondary alcs.
     is emphasized.
    ANSWER 25 OF 33 CAPLUS COPYRIGHT 2003 ACS
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1996:509667 CAPLUS

125:211144

ACCESSION NUMBER:

DOCUMENT NUMBER:

TITLE:

Complexation of LnIII and CaII Cations with 3,4-Dicarboxyinulin and Model Compounds: Methyl 3,4-Dicarboxy-.alpha.-D-fructofuranoside and 3,4-Dicarboxynystose, As Studied by Multinuclear Magnetic Resonance Spectroscopy and Potentiometry

AUTHOR(S):

SOURCE:

Johnson, Louise; Verraest, Dorine L.; Besemer, Arie C.; van Bekkum, Herman; Peters, Joop A. CORPORATE SOURCE: Laboratory of Organic Chemistry and Catalysis, Delft

University of Technology, Delft, 2628 BL, Neth. Inorganic Chemistry (1996), 35(19), 5703-5710 CODEN: INOCAJ; ISSN: 0020-1669

PUBLISHER:

American Chemical Society

DOCUMENT TYPE:

Journal

LANGUAGE:

English

Complexes of LnIII and CaII cations with 3,4-dicarboxyinulin (DCI) and model compds., Me 3,4-dicarboxy-.alpha.-D-fructofuranoside (DCF) and 3,4-dicarboxynystose (DCN) were studied using multinuclear magnetic resonance spectroscopy and potentiometric methods. Complexes of the model compds. with LnIII ions provided a feasible way in which to study complexation phenomena of the dicarboxyinulin/CaII system using NMR techniques. Information on complex geometry was derived from the effect of LnIII ions on chem. shifts and longitudinal relaxation rates. Metal-ligand stoichiometries of 1:2 and 1:1, in which the ligand coordination was tridentate as well as tetradentate, were found. Potentiometric measurements carried out with CaII yielded information on the stoichiometry as well as the cooperativity of metal ion binding by the

ANSWER 26 OF 33 CAPLUS COPYRIGHT 2003 ACS 1995:820596 CAPLUS

ACCESSION NUMBER:

ligands.

123:202726

DOCUMENT NUMBER: TITLE:

Method for oxidizing carbohydrates

INVENTOR (S):

Besemer, Arie Cornelis; de Nooy, Arjan Erik

PATENT ASSIGNEE(S):

Nederlandse Organisatie voor Toegepast-Natuurwetenschappelijk Onderzoek-, Neth.

SOURCE:

PCT Int. Appl., 20 pp.

CODEN: PIXXD2

DOCUMENT TYPE:

Patent

LANGUAGE:

English 1

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO KIND DATE APPLICATION NO. DATE WO 9507303 A1 19950316 WO 1994-NL217 19940907 W: JP, US

RW: AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE

NL 9301549 A 19950403 NL 1993-1549 19930907 20030303

NL 194919 В PRIORITY APPLN. INFO.:

NL 1993-1549 A 19930907

OTHER SOURCE(S): MARPAT 123:202726

Carbohydrates having a primary OH group, e.g., starch (I), inulin, and fractions and derivs. thereof, are oxidized by hypohalite in the presence of a catalytic amt. of di-tert-alkyl nitroxyl, esp. 2,2,6,6-tetramethylpiperidin-1-oxyl (II), in an aq. medium at pH 9-13. The catalytic amt. of nitroxyl is preferably 0.1-2.5 wt.% (based on the carbohydrate). The oxidn. affords products having a high content (>90%) of carboxyl groups, without significant chain breakdown. Thus, a 4% soln. of NaOCl was adjusted to pH 10.8 with HCl, cooled to O.degree., and added all at once to an aq. soln. contg. water-sol. potato I, II (1 wt.% based on I), and NaBr at O.degree.. The percentage uronic acid formed was a measure of the selectivity of the reaction, and anal. showed a 96% yield.

ANSWER 27 OF 33 CAPLUS COPYRIGHT 2003 ACS ACCESSION NUMBER: 1995:712425 CAPLUS

DOCUMENT NUMBER:

123:340678

TITLE:

Selective oxidation of primary alcohols

mediated by nitroxyl radical in aqueous solution.

Kinetics and mechanism

AUTHOR (S):

de Nooy, Arjan E. J.; Besemer, Arie C.; van

Bekkum, Herman

CORPORATE SOURCE:

Dep. Biochem., TNO Nutr. Food Res. Inst., Zeist, 3700

AJ, Neth.

SOURCE: Tetrahedron (1995), 51(29), 8023-32

CODEN: TETRAB; ISSN: 0040-4020

DOCUMENT TYPE: LANGUAGE:

Pergamon Journal English

The kinetics of the TEMPO-mediated oxidn. of Me AB .alpha.-D-glucopyranoside to sodium Me .alpha.-D-glucopyranosiduronate were studied. An intermediate was found which was identified as the hydrated aldehyde. This was oxidized in the same manner as the alc., with pseudo first order rate consts. ratio Kobs, ald/kobs, alc apprxeq. 7. The reaction mechanism is discussed with emphasis on steric factors and compared to literature data. Two different reaction pathways are postulated; under basic reaction conditions via a cyclic transition state and under acid reaction conditions through an acyclic transition

ANSWER 28 OF 33 CAPLUS COPYRIGHT 2003 ACS ACCESSION NUMBER: 1995:468959 CAPLUS

DOCUMENT NUMBER:

123:33535

TITLE:

Highly selective nitrosyl radical-mediated oxidation of primary alcohol groups in

water-soluble glucans AUTHOR(S):

de Nooy, Arjan E. J.; Besemer, Arie C.; van Bekkum, Herman

CORPORATE SOURCE:

TNO Nutrition and Food Research Institute, Department

of Biochemistry, Utrechtseweg 48, AJ Zeist, 3700,

SOURCE:

Carbohydrate Research (1995), 269(1), 89-98

CODEN: CRBRAT; ISSN: 0008-6215

PUBLISHER: DOCUMENT TYPE: LANGUAGE:

Elsevier Journal English

OTHER SOURCE(S):

CASREACT 123:33535

With catalytic amts. of 2,2,6,6-tetramethyl-1-piperidinyloxy (TEMPO) and hypochlorite/bromide as the regenerating oxidant in water, primary alc. groups in glucans and derivs. thereof were rapidly and completely oxidized. For pyranosides, selectivity was higher than 95% and no side products could be detected with 1H and 13C NMR or with high-performance anion-exchange chromatog. (HPAEC). The optimum pH for the reaction was between 10 and 11. The oxidn. was found to be first order in TEMPO and Br-. The oxidn. method can be applied to det. the amt. of primary alc. groups in water-sol. glucans; for pullulan, a proportion of 70% and for dextran, a proportion of 3% primary alc. groups was found.

ANSWER 29 OF 33 CAPLUS COPYRIGHT 2003 ACS ACCESSION NUMBER:

1995:177561 CAPLUS

DOCUMENT NUMBER: TITLE:

122:81806

AUTHOR (S): CORPORATE SOURCE: The hypochlorite oxidation of inulin Besemer, Arie C.; van Bekkum, Herman TNO-Nutrition, Zeist, 3700 AJ, Neth.

Recueil des Travaux Chimiques des Pays-Bas (1994),

113(9), 398-402 CODEN: RTCPA3; ISSN: 0165-0513

PUBLISHER: DOCUMENT TYPE: LANGUAGE:

Elsevier Journal English

The oxidn. of inulin with hypochlorite in the presence of sodium bromide as the catalyst has been investigated. The main objective of this study was to establish the calcium-binding properties of the product, 3,4-dicarboxy-inulin, and to optimize the reaction conditions. The polycarboxylate products are obtained in high yields (80-95%) and exhibit good to excellent calcium-sequestering properties. Typical values for the sequestering capacity (SC) of dicarboxy-inulin, prepd. from chicory-inulin with av. degree of polymn. (DP) 10, are 2.0=2.3 mmol Ca/g. Dahlia inulin (av. DP 30) yields an even better performing product (2.5 mmol Ca/g). Hence these materials are potential candidates for replacement of the often used co-builders in laundry detergent formulations, e.g. polyacrylate. The most suitable conditions for the reaction with bromide as a catalyst are found at pH 10. Without catalyst good results are obtained when the stoichiometric amt. of NaOCl is added at once to the inulin soln. and the pH is kept at 9.

ANSWER 30 OF 33 CAPLUS COPYRIGHT 2003 ACS

ACCESSION NUMBER:

1994:220725 CAPLUS

DOCUMENT NUMBER:

120:220725

TITLE:

CORPORATE SOURCE:

The catalytic effect of bromide in the hypochlorite

oxidation of linear dextrins and inulin Besemer, Arie C.; van Bekkum, Herman

TNO-Nutr., Zeist, 3700 AJ, Neth. Starch/Staerke (1994), 46(3), 101-6 CODEN: STARDD; ISSN: 0038-9056

Journal

DOCUMENT TYPE:

AUTHOR (S):

SOURCE:

LANGUAGE: English

The effect of bromide in the hypochlorite oxidn. of

amylodextrins (short chain amylose) and inulin on the reaction rate was investigated. Both in the linear dextrins and inulin oxidn., the rate increased linearly with the NaBr concn., which proved its catalytic effect. The origin of the catalysis is the rapid Br1-/OCl1conversion. The 2nd-order rate const. of the reaction carbohydrate + HOBr/OBr1- .fwdarw. polycarboxylates + HBr/Br1- was detd. at pH 9.0 and 294-298 K. The activation energy for the oxidn. of amylodextrins amts. to 67 kJ.mole-1. The highest rate was detd. at pH 8.25. The reaction rate was almost independent of pH in the region 8.5-9.5. Because of the simultaneous occurrence of Br, Br31-, hypobromous acid, and hypobromite in the pH region investigated, it was difficult to elucidate the oxidn. mechanism.

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ANSWER 31 OF 33 CAPLUS COPYRIGHT 2003 ACS
ACCESSION NUMBER:
                       1994:220724 CAPLUS
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DOCUMENT NUMBER:

120:220724

TITLE:

Dicarboxy-starch by sodium hypochlorite/bromide oxidation and its calcium-binding properties

AUTHOR (S):

Besemer, Arie C.; van Bekkum, Herman Dep. Biochem. Phys. Chem., TNO, Zeist, 3700, Neth.

CORPORATE SOURCE:

Starch/Staerke (1994), 46(3), 95-101 CODEN: STARDD; ISSN: 0038-9056

SOURCE:

DOCUMENT TYPE:

Journal

LANGUAGE:

English

Various aspects of the glycolic oxidn. of the glucose units of starch with stoichiometric amts. of NaClO using NaBr as the catalyst were investigated. The main objective of the study was the optimization of the reaction conditions for the prepn. of dicarboxy-starch and to establish the Ca sequestering capacity of the material. The products, polycarboxylates, were obtained in high yields (90-95%), and they had a sequestering capacity .ltoreq.1.4 mmole Ca/g, which was somewhat higher than that of the materials obtained in a noncatalyzed oxidn. This was attributed to the fact that the reaction with NaBr may be carried out at higher pH (>9.0) than the noncatalyzed reaction and that a relatively low concn. of NaClO could be maintained throughout the process. In this way, the decompn. of HClO/ClO1-, esp. occurring at high concn. of HClO at low pH (6.0-8.0) was prevented. Other advantages of this procedure were the colorless appearance of the materials and the higher reaction rate. The feasibility for large-scale prepn. followed from the results of a few exploratory expts., including an electrochem. prepn.

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ANSWER 32 OF 33 CAPLUS COPYRIGHT 2003 ACS
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ACCESSION NUMBER:

1991:473980 CAPLUS

DOCUMENT NUMBER:

115:73980

TITLE:

Preparation of polydicarboxysaccharides by oxidation of polysaccharides and their use as

detergent builders

INVENTOR(S):

Besemer, Arie Cornelis

PATENT ASSIGNEE(S):

Nederlandse Organisatie voor Toegepast-

Natuurwetenschappelijk Onderzoek (TNO), Neth.

SOURCE:

Eur. Pat. Appl., 11 pp. CODEN: EPXXDW

DOCUMENT TYPE:

Patent

LANGUAGE:

English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND DATE	APPLICATION NO. DATE
EP 427349	A2 19910515	EP 1990-202964 19901108
EP 427349	A3 19911113	3
EP 427349	B1 19950712	2
R: AT, BE,	CH, DE, DK, ES,	FR, GB, GR, IT, LI, LU, NL, SE
NL 8902786	A 19910603	NL 1989-2786 19891110
		NL 1990-1027 19900427
ES 2074528	T3 19950916	ES 1990-202964 19901108
CA 2029542	AA 19910511	CA 1990-2029542 19901109
CA 2029542	C 20010424	
		JP 1990-302847 19901109
	B2 20010604	
WO 9117189	A1 19911114	WO 1991-NL68 19910426
		CH, DE, DK, ES, FI, GB, HU, JP, KP, KR,
		NO, PL, RO, SD, SE, SU, US
•		CH, CM, DE, DK, ES, FR, GA, GB, GR, IT,
	MR, NL, SE, SN,	
		7 AU 1991-76960 19910426
	B2 19940512	
		EP 1991-907826 19910426
EP 526494 B1 19981209		
	_	, FR, GB, GR, IT, LI, LU, NL, SE
AI, DE,	CII, DE, DR, ES,	IR, GD, GR, II, HI, HO, NH, SE

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T2 19930930
B2 20000131
    JP 05506685
                                          JP 1991-507729 19910426
    JP 3004052
    AT 174346
                           19981215
                                          AT 1991-907826 19910426
                     E
    ES 2126570
                                          ES 1991-907826
                                                           19910426
                      Т3
                           19990401
                                          CA 1991-2081483 19910426
                      C
                           19990413
    CA 2081483
                                          NO 1992-4134
                                                           19921026
    NO 9204134
                      А
                           19921027
                                          US 1992-949498
                                                           19921207
                           19940705
    US 5326864
                      Α
                                                     A 19891110
PRIORITY APPLN. INFO.:
                                       NL 1989-2786
                                                       A 19900427
A 19910426
                                       NL 1990-1027
                                       WO 1991-NL68
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Polydicarboxysaccharides are prepd. by oxidizing polysaccharides in the presence of hypobromite and/or hypoiodite in low concn. The method gives high yields of dicarboxy product during short reaction times. The product has high Ca- and Mg-binding capacities and a white color and is useful as a phosphate substitute in detergent compns. Thus, a soln. of 7.1 g amylosedextrin in 150 mL water was treated with 0.3 g NaBr at pH 7.5-9 with 55 mL NaOCl soln. (contg. 0.10 g active Cl/mL; added at 2 mL/5 min) to prep. a dicarboxy amylosedextrin.

ANSWER 33 OF 33 CAPLUS COPYRIGHT 2003 ACS ACCESSION NUMBER: 1991:452175 CAPLUS

DOCUMENT NUMBER: 115:52175 TITLE:

Polyglucuronic acids and their preparation and use

INVENTOR(S): Kerkenaar, Antonius; Besemer, Arie Cornelis

PATENT ASSIGNEE(S): Nederlandse Organisatie voor Toegepast-

Natuurwetenschappelijk Onderzoek (TNO), Neth.

SOURCE: PCT Int. Appl., 15 pp.

CODEN: PIXXD2

DOCUMENT TYPE:

Patent

LANGUAGE:

English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 9104988	A1	19910418	WO 1990-NL142	19900928

W: JP. US

RW: AT, BE, CH, DE, DK, ES, FR, GB, IT, LU, NL, SE

19890929 NL 1989-2428 NL 8902428 A 19910416 NL 1989-2428 PRIORITY APPLN. INFO.: 19890929

Glucose-based polysaccharides such as starch and derivs. are selectively oxidized (esp. in an electrochem. cell) in the presence of complexing substances (e.g., C4-20 alkanoic acids and C3-C19 alkanesulfonic acids) to prep. polyglucuronic acids which have a high content of glucuronic acid units and are useful as complexing agents, carriers, stabilizers, soly. improvers, etc. Thus, 160 mg .beta.-cyclodextrin was oxidized in an electrochem. cell in the presence of 100 mg lauric acid to give a polyglucuronic acid.